

Acta Crystallographica Section C

**Crystal Structure
Communications**

ISSN 0108-2701

Chloro(1-{3-[2-(diphenylphosphanyl- κP)ethyl]- η^6 -benzyl]-3,5-dimethyl-1*H*-pyrazole- κN^2 })ruthenium(II) trifluoromethanesulfonate

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Electronic paper

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Chloro(1-[3-[2-(diphenylphosphanyl- κP)ethyl]- η^6 -benzyl]-3,5-dimethyl-1*H*-pyrazole- κN^2)ruthenium(II) trifluoromethanesulfonate

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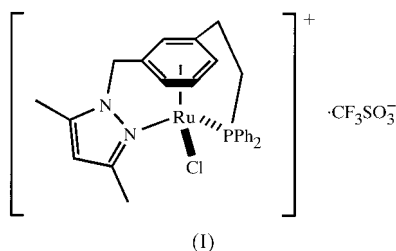
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Received 27 September 2000

Accepted 30 October 2000

Data validation number: IUC0000319

To address the question of the role of chirality at the metal in enantioselective catalysis, a pseudo-tetrahedral three-legged piano-stool complex has been prepared, *i.e.* $[\text{RuCl}(\text{C}_{26}\text{H}_{27}\text{N}_2\text{P})](\text{CF}_3\text{SO}_3)$. Anchoring a phosphine and a pyrazole tether to an arene (PArN) yields, after $\eta^6:\eta^1:\eta^1$ coordination to ruthenium, $[\{\eta^6:\eta^1:\eta^1\text{-(PArN)}\}\text{RuCl}]^+$ as a 1:1 mixture of



enantiomers. Unfortunately, all attempts to resolve the enantiomers failed. The structure solution revealed the presence of racemic crystals.

Experimental

The synthesis of the PArN(CH₃) ligand, *i.e.* 1-[3-[2-(diphenylphosphanyl)ethyl]benzyl]-3,5-dimethyl-1*H*-pyrazole, is analogous to the synthesis of PArN(CF₃) (Therrien & Ward, 1999). Replacing the bis(trifluoromethyl)pyrazole by 3,5-dimethylpyrazole affords the corresponding ligand. Only its coordination to ruthenium is fully described. To a dichloromethane solution (25 ml) of $[\text{RuCl}_2(\eta^6\text{-C}_6\text{H}_5\text{CO}_2\text{Et})_2]$ (0.129 g, 0.20 mmol), PArN(CH₃) (0.160 g, 0.40 mmol) was added. The mixture was stirred for 20 min. The volume was reduced to 10 ml and the product precipitated with hexane to afford quantitatively $[\text{RuCl}_2(\eta^6\text{-C}_6\text{H}_5\text{CO}_2\text{Et})(\eta^1\text{-PArN}(\text{CH}_3))]$. ¹H NMR (CDCl₃): 7.9–7.8 (*m*, 4H), 7.52 (*m*, 6H), 7.04 (*t*, 1H), 6.84 (*m*, 1H), 6.71 (*s*, 2H), 6.34 (*d*, 2H), 5.79 (*s*, 1H), 5.46 (*s*, 1H), 5.06 (*s*, 4H), 4.30 (*m*, 2H), 2.86 (*m*, 2H), 2.39 (*m*, 2H), 2.19 (*s*, 3H), 2.16 (*s*, 3H), 1.34 p.p.m. (*t*, 3H). ³¹P{¹H} NMR (CDCl₃): 23.2 p.p.m. Into a 25 ml pressure Schlenk flask was introduced $[\text{RuCl}_2(\eta^6\text{-C}_6\text{H}_5\text{CO}_2\text{Et})(\eta^1\text{-PArN}(\text{CH}_3))]$ (0.180 g, 0.25 mmol) in dichloromethane (10 ml). After

three freeze–pump–thaw cycles, the solution was heated at 393 K for 48 h, cooled to room temperature and the product precipitated with hexane (200 ml). The yellow–orange solid was filtered off and washed with ether. Purification by flash chromatography with CH₂Cl₂–MeOH (20:1), yielded $[\text{RuCl}_2\{\eta^6:\eta^1\text{-PArN}(\text{CH}_3)\}]$ (0.038 g, 0.07 mmol, 27%). ¹H NMR (CDCl₃): 7.82 (*m*, 2H), 7.64 (*m*, 2H), 7.39 (*m*, 6H), 5.89 (*dd*, 1H), 5.85 (*s*, 1H), 5.71 (*d*, 1H), 5.36 (*d*, 1H), 5.16 (*d*, 1H), 5.03 (*d*, 1H), 4.68 (*s*, 1H), 3.49 (*m*, 2H), 2.53 (*m*, 2H), 1.64 (*s*, 3H), 1.27 p.p.m. (*s*, 3H). ³¹P{¹H} NMR (CDCl₃): 46.2 p.p.m. To a CHCl₃ solution (5 ml) of $[\text{RuCl}_2\{\eta^6:\eta^1\text{-PArN}(\text{CH}_3)\}]$ (300 mg, 0.51 mmol), AgOSO₂CF₃ (180 mg, 0.51 mmol) was added. The mixture was stirred at room temperature for 24 h. After filtration through Celite, the solution was evaporated to dryness to afford quantitatively $[\text{RuCl}\{\eta^6:\eta^1\text{-PArN}(\text{CH}_3)\}(\text{OSO}_2\text{CF}_3)]$. After a week at room temperature in CHCl₃, crystals suitable for X-ray analysis were obtained. ¹H NMR (CDCl₃): 8.03 (*m*, 2H), 7.7–7.1 (*m*, 8H), 6.79 (*s*, 1H), 6.33 (*br*, 1H), 6.21 (*d*, 1H), 5.66 (*s*, 1H), 5.53 (*d*, 1H), 5.18 (*d*, 1H), 4.69 (*d*, 1H), 3.81 (*m*, 1H), 3.48 (*m*, 1H), 3.18 (*m*, 1H), 2.91 (*m*, 1H), 1.44 (*s*, 3H), 1.26 p.p.m. (*s*, 3H). ³¹P{¹H} NMR (CDCl₃): 47.4 p.p.m.

Crystal data

$[\text{RuCl}(\text{C}_{26}\text{H}_{27}\text{N}_2\text{P})](\text{CF}_3\text{O}_3\text{S})$
 $M_r = 684.06$
 Triclinic, $P\bar{1}$
 $a = 9.3674$ (3) Å
 $b = 11.0111$ (3) Å
 $c = 13.6724$ (4) Å
 $\alpha = 94.198$ (1)°
 $\beta = 92.139$ (1)°
 $\gamma = 98.968$ (1)°
 $V = 1387.57$ (7) Å³

$Z = 2$
 $D_x = 1.637$ Mg m⁻³
 Mo $K\alpha$ radiation
 Cell parameters from 3988 reflections
 $\theta = 1.50$ – 26.39 °
 $\mu = 0.847$ mm⁻¹
 $T = 293$ (2) K
 Plate, orange
 $0.42 \times 0.11 \times 0.08$ mm

Data collection

Siemens SMART CCD diffractometer
 ω scans, 1271 frames, 0.30°, 20 s,
 detector distance 5.5 cm, detector
 angle 23.0°
 6978 measured reflections
 5008 independent reflections

4074 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.039$
 $\theta_{\text{max}} = 26.39$ °
 $h = -11 \rightarrow 11$
 $k = -13 \rightarrow 7$
 $l = -16 \rightarrow 15$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.051$
 $wR(F^2) = 0.120$
 $S = 1.183$
 5008 reflections
 356 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0109P)^2 + 4.6884P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.54$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.84$ e Å⁻³

Data collection: SMART (Siemens 1994–1996); cell refinement: SAINT (Siemens 1994–1996); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); software used to prepare material for publication: SHELXL97.

This work was supported by the Swiss National Science Foundation and the Stiftung für Stipendien auf dem Gebiete der Chemie (through the award of an A. Werner Fellowship to TRW, 1994–1999).

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